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## Phosphorus, Sulfur, and Silicon and the Related Elements

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## Structure of Low-Coordinated Phosphorus (III) Compounds Including Phosphorus-Nitrogen Multiple Bond: <sup>13</sup>C, <sup>15</sup>N and <sup>31</sup>P NMR Investigation

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STRUCTURE OF LOW-COORDINATED PHOSPHORUS (III) COMPOUNDS INCLUDING PHOSPHORUS-NITROGEN MULTIPLE BOND: 13C. 15N AND 31P NMR INVESTIGATION

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A wide series of low-coordinated phosphorus (III) compounds including iminophosphines 1, two- 2 or mono-coordinated 3 phosphorus-containing cations, and anion 4, where  $Mes^* = 2,4,6$ -tris-(tert-butyl)phenyl, has been investigated by  $^{13}C$ ,  $^{15}N$  and  $^{31}P$  NMR spectroscopy. The spectral parameters obtained are considered in connection with the ab initio (6-31G\*) calculation data of the model compounds.

Mes\*-N=P-X Mes\*-NX-P+-NR<sub>2</sub> Mes\*-N=P+ Mes\*-N=P-=NR Me<sub>4</sub>N+ 
$$\frac{1}{1}$$
  $\frac{2}{2}$   $\frac{3}{4}$   $\frac{1}{1}$  X= Alk, Ar, NR<sub>2</sub>, OR, SR, PR<sub>2</sub>, Hal;  $\frac{2}{1}$  X= H, AlCl<sub>3</sub>-, GaCl<sub>3</sub>-

In spite of the absence of the conjugation between the P=N and Mes\*  $\pi$ -systems, essential  $^{13}$ C shielding variation is found for all carbon-13 nuclei in Mes\* moiety within the series 1-4. The  $\pi$ -inductive nature of the  $^{13}$ C shielding variation in the aromatic moiety is suggested. The found peculiarity of the Mes\* substituent to be polarized by electronegative substituents seems to be an additional factor promoting the total stability of low-coordinated phosphorus (III) compounds involving the Mes\* moiety. The  $^{13}$ C NMR data analysis makes it possible to determine the isomers configuration for compounds 1,2.

The <sup>15</sup>N and <sup>31</sup>P NMR chemical shifts of corresponding nuclei in Mes\*-N=P-X moiety, as well as phosphorus-nitrogen coupling constants, are shown to be sensitive both to the hybridization type of the P and N atoms and to inductive and resonance properties of the X substituent. The largest deshielding of <sup>15</sup>N nuclei (δN=+163.6 ppm) is found for anion 4. The P=N bonds have the double character in this compound but they are highly polarized to N atoms.

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